

## Arm & Hammer<sup>™</sup> Sodium Bicarbonate USP No. 1 Powdered

Reviewed: April 1, 2017	Test Method	USP	FCC
Description	-	Sodium Bicarbonate contains not less than 99.0 percent and not more than 100.5 percent of NaHCO <sub>3</sub> calculated on the dried basis.	A white crystalline powder. It is stable in dry air, but slowly decomposes in moist air. Its solutions, when freshly prepared with cold water, without shaking, are alkaline to litmus. The alkalinity increases as the solutions stand, are agitated or are heated.
Assay – dry basis	USP	Not less than 99.0% and not more than 100.5% of NaHCO <sub>3</sub>	Not less than 99% NaHCO <sub>3</sub> after drying
Identification	USP <191>	Meets the requirements of the tests for sodium and bicarbonate.	A 1 in 10 solution gives positive tests for sodium and for bicarbonate.
Insoluble Substances	USP	Dissolve 1 g in 20 ml of water; the resulting solution is complete and clear.	Passes test
Normal Carbonate	USP	Meets test.	-
Chloride	USP <221>	Not more than 0.015%	-
Limit of Sulfur Compounds	USP	Not more than 0.015%	-
Elemental Impurities	ICP		-
Cadmium		Not more than 0.5 µg/g	-
Lead		Not more than 0.5 µg/g	Not more than 2 mg/Kg
Arsenic		Not more than 1.5 µg/g	-
Mercury		Not more than 1 µg/g	-
Limit of Ammonia	NA – See remarks	Not more than 20 ppm	-
Loss on Drying	USP <731>	Not more than 0.25%	Not more than 0.25% by weight
the product insure that ammoni *Elemental Impurities (replaces Table 2 elements are not known Residual Solvents testing under	a will not exceed the USP Heavy Metals <231>)Limi n or expected impurities. r USP <467> is not require	urch & Dwight Sodium bicarbonate. Con limit. Absence of ammonia is confirmed ts based on USP <232> Table 2, Oral D ed as no solvents, and specifically no sol cation of Church & Dwight Sodium Bicar	on each lot via olfactory test. rug Products. The remaining vents of Class 1, 2, 3 or Table 4

## Granulation (Powdered)

		Ro-Tap Cumulative % Retained	
Sieve Size (USS)	Microns	Minimum	Maximum
100	149	0	2
200	74	20	45
325	44	60	100

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General Properties (Not Specifications)

Empirical Formula	NaHCO <sub>3</sub>	
CAS Number	144-55-8	
Other Names	Bicarbonate of Soda	
	Sodium Hydrogen Carbonate	
	Baking Soda	
Chemical Abstract Name	Carbonic acid monosodium salt	
E Number	E-500(ii)	
Appearance White crystalline powder		
aste Slightly alkaline		
Molecular Weight	84.01	
Thermal Decomposition	Decomposes without melting into Na <sub>2</sub> CO <sub>3</sub> , H <sub>2</sub> O and CO <sub>2</sub> .	
Crystal Density	137.3 lb /ft³, 2.2 g / cc	
Bulk Density	63 lb/ft <sup>3</sup> , 1.009 g/cc	
BTU / lb at 72°F	0.249	
Solubility in water at 77°F	Approximately 9.5%	
Solubility in Alcohol	Insoluble	
Alkali Equivalent	1 lb NaHCO <sub>3</sub> = 0.369 lb Na <sub>2</sub> O	
Acid Equivalent	1 lb NaHCO <sub>3</sub> = 0.435 lb HCl	
Carbon Dioxide Equivalent	1 lb NaHCO <sub>3</sub> = 0.524 lb CO <sub>2</sub>	
pH 1% aqueous soln at 77°F	Approximately 8.3.	

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